

PHOTOCATALYTIC APPLICATIONS OF ELECTROSPUN TiO₂ NANOFIBERS PREPARED FROM POLY(BUTYLENES ADIPATE-COTEREPHTHALATE)/NANOCOMPOSITES.

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Titanium dioxide (TiO₂) is one of the more used semiconductors available for photocatalysis, due to its high photoactivity, low cost, chemical and biological inertness and durability. In photocatalytic reactions a high surface-to-volume ratio (S/V) is extremely important. Nanoparticles (NPs) satisfy that criterion, but in normal applications TiO₂ NPs have a strong tendency to agglomerate into larger particles. TiO₂-based fibers with large aspect ratio can overcome the agglomeration problem keeping high S/V ratio. In the present contribution TiO₂ nanofibers (NFs) were prepared using the electrospinning technique. In a first step nanocomposites fibers were produced by dispersions of TiO₂ NPs (AEROXIDE®TiO₂-P25) on a synthetic biodegradable aliphatic-aromatic copolyesterpoly(butylenesadipate-coterephthalate) (PBAT) using 2,2,2 Trifluorethanol as solvent. The TiO₂ concentration was varied as: 0.04%, 2.5%, 5%, 10% and 12.5% w/w. In a typical electrospinning procedure, the precursor solution was loaded into a syringe equipped with a stainless steel needle. An electric voltage of 16.5 kV was applied between the needle and the target. The distance between the needle and the target was 15 cm and a stable flow rate of 1 mL/h was maintained during the process. After the electrospinning step the fibers were thermal threaded (TT) at 450°C in air atmosphere to eliminate de organic contents of the composite and keep the more active photocatalytic anatase phase intact. The fibers were analyzed before and after TT by water contact angle, optical profilometry, SEM, EDS, FTIR-ATR and DRX. Specific surface area was measured by BET. The results showed the formation of nanocomposite fibers of 250-1000 nm in diameter. After TT the TiO₂ nanofiber dispersion was maintained. Some degree of agglomeration was observed when a high TiO₂ concentration was used. Ti elemental concentration measured by EDS matched the initial concentration used for the NFs preparation. For example, when the TiO₂ concentration used in the initial mixture was 0.04%, 2.5%, 5% and 10% w/w, the elemental composition of Ti measured by EDS was 0.03%, 2.5%, 5.2% and 11.3% respectively. FTIR-ATR results showed an increase in the signal corresponding to the broad band at 700 cm⁻¹ corresponding to Ti-O-Ti stretching of TiO₂ when the TiO₂ concentration used in the initial dispersion increased. EDS and FTIR-ATR results confirmed the efficient incorporation of TiO₂ in the prepared nanocomposites fibers. Preliminary tests of the photocatalytic activity of TiO₂ NFs for H₂ generation using methanol/water solutions (10% v/v) and 12.5% TiO₂/PBAT NFs showed a higher rate of H₂ evolution for the NFs than for the TiO₂ NPs; in spite the specific area was similar for NFs and NPs. Photocatalytic tests for degradation of pollutants using methyl orange as a prototype compound are under way.

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